Accepted Manuscript

Title: Development and validation of a multi-residue method for pesticide determination in honey using on-column liquid-liquid extraction and liquid chromatography-tandem mass spectrometry

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PII:	S0021-9673(07)00502-X
DOI:	doi:10.1016/j.chroma.2007.03.035
Reference:	CHROMA 347516
To appear in:	Journal of Chromatography A
Received date:	25-10-2006
Revised date:	28-2-2007
Accepted date:	9-3-2007

Please cite this article as: C. Pirard, J. Widart, B.K. Nguyen, C. Deleuze, L. Heudt, E. Haubruge, E. De Pauw, J.-F. Focant, Development and validation of a multi-residue method for pesticide determination in honey using on-column liquid-liquid extraction and liquid chromatography-tandem mass spectrometry, *Journal of Chromatography A* (2007), doi:10.1016/j.chroma.2007.03.035

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S	ubmitted for publication in a special issue of Journal of Chromatography A
	ADV SAMPLE PREP 2006
	Guest Editor: Dr L. Ramos
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45 Abstract

We report on the development and validation under ISO 17025 criteria of a multi-residue 46 confirmatory method to identify and quantify 17 widely chemically different pesticides 47 (insecticides: Carbofuran, Methiocarb, Pirimicarb, Dimethoate, Fipronil, Imidacloprid; 48 49 herbicides: Amidosulfuron, Rimsulfuron, Atrazine, Simazine, Chloroturon, Linuron, Isoxaflutole, Metosulam; fungicides: Diethofencarb) and 2 metabolites (Methiocarb sulfoxide 50 51 and 2-Hydroxytertbutylazine) in honey. This method is based on an on-column liquid-liquid extraction (OCLLE) using diatomaceous earth as inert solid support, and liquid 52 53 chromatography (LC) coupled to mass spectrometry (MS) operating in tandem mode 54 (MS/MS). Method specificity is ensured by checking retention time and theoretical ratio 55 between two transitions from a single precursor ion. Linearity is demonstrated all along the range of concentration that was investigated, from 0.1 to 20 ng g^{-1} raw honey, with correlation 56 57 coefficients ranging from 0.921 to 0.999, depending on chemicals. Recovery rates obtained on home-made quality control samples are between 71 and 90%, well above the range defined by 58 the EC/657/2002 document, but in the range we had fixed to ensure proper quantification, as 59 levels found in real samples could not be corrected for recovery rates. Reproducibility was 60 found to be between 8 and 27%. Calculated CC α and CC β (0.0002-0.943 ng g⁻¹ for CC α , and 61 0.0002-1.232 ng g⁻¹ for CC β) show the good sensitivity attained by this multi-residue 62 63 analytical method. The robustness of the method has been tested in analysing more than 100 64 raw honey samples collected in different areas in Belgium, as well as in some wax and bee 65 samples with a slightly adapted procedure.

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Key words: Pesticides, liquid chromatography, tandem mass spectrometry, honey,
diatomaceous earth, on-column liquid-liquid extraction

70

70 1. INTRODUCTION

During recent years, some beekeepers were confronted to hush of their beehives in Belgium. 71 As external observers did not find any classical illness in those hives, several hypotheses were 72 73 proposed to explain the withering: specific illnesses, pesticide contamination, poor beekeepers 74 practices, etc. Although, several projects were conducted in Europe for the study of separated factors such as imidacloprid levels, no satisfactory explanation could however be given. A 75 76 multifactorial study has thus been initiated at the Belgian Walloon Region level. Specialized 77 observers have studied beekeeping practices, common illness in the hives and environmental 78 conditions around hives for selected Belgian locations. Consequently, a list of pesticides 79 including both product types used in apiculture and the surrounding agriculture has been 80 extracted. This list consists in a large number of compounds belonging to different chemical classes of insecticides, herbicides and fungicides, which could be the source of the bee 81 82 decline. In order to verify this assumption, a wide range of pesticides has to be monitored in 83 different honey, wax and bee samples coming from different injured and safe areas.

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In the past decades numerous publications have already reported analytical method for 85 86 pesticides determination in honey, and a review published four years ago has highlighted the 87 most relevant ones [1]. These studies however usually focused on the analysis very few 88 compounds, often belonging to one or two pesticide families at the most, such as 89 organochlorine or organophosphorous residues. As demands of pesticide analysis started to be 90 more motivated according similar agrarian uses rather than similar physico-chemical 91 properties of chemicals, multi-analyte determinations have appeared since the end of nineties, 92 covering several classes of pesticides [2-5]. The challenge of such strategy is to include in a 93 single procedure a broad range of compounds having widely different structures and 94 properties.

95 Among the different reported analytical procedures, liquid-liquid extraction (LLE) is the most popular technique for such difficult matrix as honey [1, 6-8]. However, LLE requires large 96 97 amounts of solvent, is time consuming, laborious and not well suited for automation [1, 6]. As 98 an alternative, solid phase extraction (SPE) or matrix solid phase dispersion (MSPD) have 99 been widely developed in the past decades. Their simplicity, robustness, rapidity and low 100 solvent consumption are attractive parameters for the analytical chemist. Whereas SPE is 101 based on the retention of selected analytes on cartridge sorbents and their elution with 102 appropriate solvent, MSPD consists in the dispersion of the matrix on a free-adsorbent and its 103 homogeneous packing on a column prior to elution of compounds with organic solvent allowing the extraction of semi-solid and solid samples [9, 10]. The other side of the coin is 104 105 its poor capabilities for high sample input [11]. Solid-phase microextraction (SPME) has also been studied for pesticide analysis in honey [2, 12] but showed sample input limitations and 106 107 relatively high limit of detection [11]. Supercritical fluid extraction (SFE) and stir-bar 108 sorptive extraction (SBSE) still remains quite marginal in this area until now [5, 13]. In this 109 study, an on-column liquid-liquid extraction (OCLLE) method has been tested as it seemed to 110 combine advantages of LLE, SPE and SPME.

111

112 Mass spectrometery represents the most selective detector for pesticides as it provides 113 structural information allowing unequivocal confirmation and its use in a multi-residue 114 screening context. Although GC is often reported as the most powerful separation tool, it 115 involves a derivatization step for thermally unstable compounds. This introduces additional 116 handling and reaction, thus potentially reducing reproducibility and recovery rates [1].

The goals of the present study were to develop and validate under ISO 17025 criteria a multiresidue screening method to identify and quantify 17 widely chemically different insecticides (Carbofuran, Methiocarb, Pirimicarb, Dimethoate, Fipronil, Imidacloprid), herbicides

120 (Amidosulfuron, Rimsulfuron, Atrazine, Simazine, Chlorotoluron, Linuron, Isoxaflutole, Metosulam), fungicides (Diethofencarb) and some metabolites (Methiocarb sulfoxide and 2-121 122 Hydroxytertbutylazine) potentially present in honey. This method is based on on-column liquid-liquid extraction (OCLLE) and liquid chromatography coupled to mass spectrometry 123 (LC-MS) operating in tandem mass spectrometry mode (MS/MS). This analytical procedure 124 125 was evaluated according to European Commission advice 2002/657/EC [14] in terms of trueness, reproducibility, sensitivity, specificity and robustness. A slightly adapted procedure 126 127 was also developed and applied to wax and bee samples.

128

129 **2. EXPERIMENTAL**

130 2.1. Reagents and standards

Water was obtained from a Milli-Q[®] Ultrapure Water Purification Systems (Millipore, 131 Brussels, Belgium). Acetone, ethyl acetate, and acetonitrile were Pestanal[®] reagents 132 133 (Promochem, Molsheim, France) while NaCl was from Acros Organics (Geel, Belgium) and 134 acetic acid from JT Baker (Deventer, The Netherlands). Chem Elut cartridges (5 mL) were purchased from Varian Inc. (Varian, Sint-Katelijne-Waver, Belgium). These disposable 135 136 cartridges contain cleaned diatomaceous earth packed in pure polypropylene housing and a 137 hydrophobic membrane at the base of the cartridge to ensure that moisture is excluded from 138 the extract. Liquid nitrogen was purchased at Air Liquide (Liege, Belgium). All pesticide 139 reference standards are produced by Dr. Ehrenstorfer (Augsburg, Germany), their 140 concentrations are listed in Table 1. Linuron D6 (C₉H₄Cl₂N₂O₂D₆) also produced by Dr. Ehrenstorfer (100 μ g mL⁻¹) was used as deuteriated surrogate standard to check extraction 141 142 step because of it easy commercial availability and its suitable retention time on the LC 143 column. PALL Bulk GHP Acrodic 13 mm syringe filters (pore diameter: 0.2 µm) were 144 purchased from VWR International Belgique (Leuven, Belgium)

145 [INSERT TABLE 1 ABOUT HERE]

Honey, wax and bee samples were collected in 16 hives spread out in Wallonia (Belgium)during the winter 2004-2005.

148 **2.2. Sample preparation**

149 Extraction of pesticides in honey was inspired by the procedure developed by Klein *et al.* [15] 150 for multi-residue determination in fruits and vegetables. It has been modified to extract the 151 selected chemicals from raw honey samples. Aliquots of 1 g of honey were spiked with 20 µL 152 of surrogate standard (see Table 1 for concentrations) before mechanical transversal agitation 153 with 1.25 ml of water and 2.5 mL of acetone for 1 hour. A 20% NaCl solution (1.25 mL) was 154 then added and the mix was loaded on the Chem Elut cartridge. After a waiting period of 15 155 min, analytes were eluted by gravity twice with 10 mL of ethyl acetate. Extracts were then 156 evaporated at 30°C until dryness under a gentle stream of nitrogen, and transferred with 200 157 μ L of an acetonitrile-water solution (10:90) in vials suited for LC injection.

158 Classical LLE was performed as follow: 6.5 mL of acetonitrile were added to 1 g of honey 159 dissolved in 2 mL of water, and mechanically shaken for 30 min. Organic and aqueous phases 160 were separated by centrifugation (15 min at 2000 rpm). Organic layer was then evaporated 161 down to 100 μ L, and added to 100 μ L of water. This final extract is filtered before being 162 injected in LC-MS.

Bee samples were frozen with liquid nitrogen and crushed to obtain a fine homogeneous powder. OCLL extraction was then performed on aliquots of 0.5 g with the same procedure than described for honey.

Wax extraction was quite different. Samples were also frozen and ground in a fine powder. 0.5g was weighted, spiked with 20 μ L of surrogate standard and agitated with 10 mL of hexane and 10 mL of acetonitrile for 1 hour. Centrifugation was applied for 20 minutes and the organic phase was re-extracted with 10 mL of acetonitrile. Both aqueous phases were

170 evaporated at 30°C until dryness under a gentle stream of nitrogen. 200 µL of an acetonitrile-

171 water solution (10:90) were added, filtered and transferred in vials suited for LC injection.

172 **2.3. Instrumental analysis**

Analysis were carried out on a Quattro Ultima Platinum triple quadrupole mass spectrometer 173 174 coupled to an Alliance 2690 liquid chromatograph (Waters, Manchester, UK). The 175 chromatograph was equipped with a Polaris C18-A HPLC column (150 mm x 2.0 mm, 3 µm, 176 200 Å) from Varian Inc. (Varian, Sint-Katelijne-Waver, Belgium), kept at 40°C. The mobile 177 phase consisted of acetonitrile and water, both acidified with 0.1% of acetic acid. Gradient was applied at a flow rate of 0.4 mL min⁻¹ as follow: held the initial conditions of 10% 178 179 acetonitrile in water for 1 min, increased linearly to 80% in 14 min, increased linearly to 180 100% in 2 min, held at 100% during 1.9 min, returned to initial conditions in 0.1 min and 181 maintained for 4 min. The LC effluent was split using a T-splitter to produce a flow of 0.2 mL min⁻¹. The quadrupole mass spectrometer was equipped with a Z-spray source for positive 182 183 electrospray ionization (ESI). Capillary and cone voltages were set respectively at 3 kV and 184 35 V, temperature source was kept at 125°C while desolvatation temperature was held at 185 250°C. Nitrogen was used as cone and desolvating gas at a flow rate of 100 and 680 L/h 186 respectively. Mass spectrometer operated in MS/MS mode using multiple reactions monitoring (MRM). 99.8% pure argon from Air liquide (Liège, Belgium) was used as 187 collision gas at a constant pressure of $2x10^{-3}$ mbar. Table 2 summarises the acquisition 188 189 window definition, masses of parent and daughter ions that are monitored, and the optimized 190 collision induced dissociation (CID) voltages.

191 [INSERT TABLE 2 ABOUT HERE]

192

192 **3. RESULTS AND DISCUSSION**

3.1. Extraction assays

194 Pesticide extractions from raw honey were carried out by on column liquid-liquid extraction. 195 This technique is based on classical LLE principle, but assisted by inert solid support. This 196 inert matrix consisted in diatomaceous earth, well-known for its high porosity, high dispersing 197 capacities, and its high capacity for aqueous adsorption [16, 17]. It has been already used in 198 several chemical extraction applications [18-22]. In the present extraction step, diatomaceous 199 earth has been factory pre-packed in disposable cartridges commercially available under trade 200 name of Chem Elut, from Varian Inc. Comparison between OCLLE and classical LLE has 201 been carried out to check extraction efficiency and suitability of the procedure. Results are 202 shown in Figure 1. OCLLE seemed to provide similar or even higher extraction efficiency 203 and higher repeatability than LLE for some compounds. Moreover and above all, OCLLE 204 provides the real advantage over LLE to avoid emulsion formation in ensuring immiscibility 205 of organic solvents and aqueous matrix. This therefore significantly eases extraction 206 procedure [23]. Fidente *et al.* [24] have developed an extraction procedure based on identical 207 principles for insecticide analysis in honey but this study involved a single class of 208 insecticides and therefore included a limited number of chemically related compounds. 209 Moreover, cartridges that were used appeared to require a drying step using nitrogen flow, 210 increasing the analysis time and the procedure complexity. The strong points of Chem Elut 211 cartridges were their ease of use and the wide range of compounds that could be extracted 212 efficiently.

213 [INSERT FIGURE 1 ABOUT HERE]

214

215 **3.2. LC-MS/MS data**

216 Most of target pesticides are separated by LC prior to MS detection except Dimethoate and 217 Fipronil, which are reported to be analyzed by GC-MS or SPME-GC-MS. Avoid 218 derivatization step and allow a less rugged clean-up [6] were the reasons which led us to use 219 liquid instead of gas chromatography. The configuration of the Z-spray source designed at 220 first to prevent fragmentation during ionization [25] enhances this robustness in terms of 221 matrix related interferences, as only charged species enter in the detector. The use of tandem 222 mass spectrometry confers high specificity and reduces the risk of potential interferences 223 related to the complexity of the matrix. Each precursor ion was fragmented by Collision 224 Induced Dissociation (CID) and the two most abundant produced ions were monitored. In 225 addition to this gain of selectivity, the use of the MS/MS mode substantially increases 226 sensitivity by limiting the high background noise related to the honey matrix.

227 The Polaris C18-A column is normally dedicated to drug and drug metabolite discovery [26, 228 27]. The silica phase of this HPLC column is bonded to octadecyl chain with a polar group 229 maximizing polar retention and selectivity, and eliminating silanol residues. This allowed to 230 cover a broad range of chemically different compounds. LC gradient has been optimized to 231 distinguish the 17 pesticides keeping in mind that coeluted compounds showing different 232 masses could be separated by the mass spectrometer using multiple reaction monitoring 233 (MRM) mode. In order to achieve the best compromise between time analysis and sensitivity, 234 the number of transitions in a single window has been limited to 12. As for each precursor 235 compounds, two product ions have been recorded, this represented a maximum of 6 pesticides 236 monitored by acquisition window. An example of chromatogram is presented in Figure 2 for a 237 methanolic standard solution showing pesticide concentrations ranging between 0.4 and 20 ng mL⁻¹ depending on the congener (Tabe 1). Total LC cycling (separation and return to start 238 239 conditions) program was 23 min.

240 [INSERT FIGURE 2 ABOUT HERE]

241

242 **3.3. Method validation**

243 Specific guidelines have been produced by the European Commission for the validation of both vegetal and animal product analysis [14, 28]. As honey is product of animal origin such 244 245 as eggs or milk by-products, the analytical procedure has been validated in compliance with 246 the European Commission decision EC/657/2002 [14] dedicated to the measurement of 247 residues in living animals or their derived products. This group of contaminants includes 248 banned compounds or residues for which maximum levels (MRL) have been assigned 249 (compounds belonging to the group B from annex 1 of 96/23/EC [29]). Although pesticides 250 targeted in this work did not belong to this "blacklist", this guideline has been chosen because 251 of its particular rigor and precision. Validation will therefore involve determination of the 252 specificity, the calibration curves, the trueness, the accuracy (repeatability and 253 reproducibility), the sensitivity and the robustness.

254

255

255 3.3.1. Specificity

256 In order to prevent misidentification of analytes due to interferences, relative retention time 257 (RRT) has been checked for each pesticides and a maximum deviation of 2.5% from the 258 expected RRT accepted. Additionally, two transitions from a single precursor ion were 259 monitored to complete identification insurance. These transitions have been chosen for each 260 target on standard solution as both most abundant ions produced from precursor. Figure 3 shows an example for the determination of most intense fragments obtained by MS/MS for 261 262 Imidacloprid and optimisation of collision voltages. Identification of analytes was confirmed 263 if isotopic ratio bias from standard theoretical ratios were below 20%. Although two produced ions have been recorded, quantification has been performed using only one mass because of 264 265 software limitation. These masses represent the most intense ions produced and are listed in 266 bold underlined in Table 2.

- 267 [INSERT FIGURE 3 ABOUT HERE]
- 268

269 3.3.2. Calibration curves

Calibration curves have been produced for quantification. They were built using blank honey matrix spiked after the extraction step at 5 different pesticide levels, the zero point included (Table 3). This calibration procedure permits to avoid matrix effect in the electrospray source, such as ion enhancement or suppression. Additionally to criteria required by the 2002/657/EC, a second calibration curve was run at the end of each sample series to check the stability of the detector answer after unknown sample data aquisition. The requirement we set was that the end curve had to show a bias lower than 20%, compared to initial calibration.

277 [INSERT TABLE 3 ABOUT HERE]

Linearity has been observed all along the area of concentration studied depending on thechemicals. These ranges of concentrations were selected in function of the sensitivity of the

280 mass spectrometer towards each pesticide. They are listed in Table 3, together with correlation coefficient (R²) of the linear regression. Very few compounds showed residual 281 282 level or background signal in the unfortified honey matrix. A chromatogram built with 283 specific masses of these compounds and the mass of deuteriated Linuron (D6) for comparison 284 is shown in Figure 2. This background noise was very low and usually non-significant. As 285 these traces were already taken into account during the calibration, no correction by mean of 286 subtracting blank matrix levels was necessary nor applied during the quantification process. 287 However, 10 blank matrices were run during the validation to ensure a minimal risk of 288 interferences and guarantee specificity of the method. Additionally, a blank matrix sample 289 was added to each unknown sample series in order to check for lab and solvent potential 290 contamination.

291

292 3.3.3. Recovery (Trueness)

Other pools of blank honey have been fortified prior to the extraction step for home-made quality control samples (QC) at 3 different levels, reported in Table 3. For each level, 6 QC samples have been run. Recoveries have been calculated as the ratio between levels measured in the QC and amounts really added to these blank samples. Particular care has to be taken for the evaporation step. Most pesticides appeared to be really sensitive to dry evaporation, and recoveries can be cut by more than a half if compounds remain to dryness a too long period even at a maximum of 30°C.

Recoveries and relative standard deviation (RSD) are listed in Table 3. According to the 2002/657/EC document, these recoveries have to range between 50 and 120%. As already mentioned, as levels found in real samples could not be corrected by the recovery rates, a narrower range, between 70 and 110 %, was chosen to ensure at best proper quantification. Although target analytes were characterized by significantly different physico-chemical

305 properties (functional groups, polarity), recorded recoveries were constantly high, accounting 306 for the versatility and efficiency of the extraction and detection methods. One can also 307 mention that if one would use the less strict 50 to 120% acceptation range stated in the 308 European Directive, one could extend the list of pesticides adequately analyzed by this 309 procedure to other chemicals such as Tribenuron-methyl, Bitertanol, Difenoconazole, 310 Flusilazole, Difenoconazole, Metazachlor, Trifloxystrobin, Metconazole and, importantly, the 311 widely spread Rotenone.

312

313 3.3.4. Repeatability and reproducibility

According to the 2002/657/EC document, 3 different QC levels have to be analyzed with six 314 315 replicate for each level, and these have to be performed on 3 distinct days in order to calculate 316 the method repeatability, as the standard deviation (SD) of the recovery mean. 317 Reproducibility has to be evaluated similarly with minor changes, such as with different 318 operators, different environment, with different solvent batches, etc. In this study, only 2 319 different QC levels (#1 and #3 in Table 3) were used to keep the validation cost in the budget, 320 resulting in a total of 36 QC measurements. Because different operators contributed to these 321 validation tests, only reproducibility was gathered in Table 3. RSD ranging between 8 and 322 27% were judged satisfactory regarding the low levels we dealt with in this work.

323

324 3.3.5. Decision limits and detection capacity (CCa and CCß)

Two different methods can be used to evaluate the decision limit (CC α) when there is no maximal residue limits (MRL) applied for the target pesticides. The first one consists in the analysis of 20 blank materials. The CC α is then equal to three times the signal-to-noise ratio (S/N) in the chromatogram where the analyte is expected. As very few compounds have shown background noise in the time window where they show up, this calculation approach

330 appeared not to be appropriate to properly evaluate the decision limit. The other method is based on the analysis of blank honey matrices spiked prior to extraction with decreasing 331 332 amounts of compounds, and the comparison between recorded MS signals and concentrations added. $CC\alpha$ is then equal to the concentration corresponding to the sum of the intercept of the 333 linear regression and the reproducibility multiplied by 1.64 ($\alpha = 5\%$ as stipulated for 334 335 compounds belonging to Group B from annex 1 of EU Directive 96/23/EC). Decision limit 336 values obtained for all investigated compounds are listed in Table 4. Detection capacities 337 $(CC\beta)$ have been calculated as the concentration corresponding to CCa added to the 338 reproducibility multiplied by a factor of 1.64 ($\beta = 5\%$). Those values are also listed in Table 4.

339 [INSERT TABLE 4 ABOUT HERE]

Because very few authors have already reported such validation data on honey and, as different criteria were applied when they did so, comparison with other reported methods is difficult. Our procedure nevertheless demonstrates to offer very good sensitivity compared to limits of detection (LODs) and quantification (LOQs) reported for some pesticides in honey by Albero *et al.* [4] and Fidente *et al.* [24].

345

346 3.3.6. Robustness

109 raw honey samples collected in different areas of Belgium have been analysed within the 347 348 scope of the multifactorial study described in the introduction. In practice, honey samples 349 appeared to be characterized by different appearances, depending on the area where they were 350 collected, with colour ranging from light yellow to dark brown, with different viscosity, etc. 351 Table 3 presents the number of samples in which target pesticides have been found 352 (considered as positive) and ranges of levels measured. Some pesticides which had not 353 successfully passed the validation criteria but had nevertheless showed recovery rates ranging 354 between 50 and 70% have been included in that Table (i.e. Rotenone, Bitertanol and

355 Flusilasole). For those, CC α and CC β have been calculated using the standard deviation 356 evaluated on 18 QCs instead of 36 QCs.

357

358 Additionally to honey samples, some bee (99) and wax (98) samples have also been analyzed 359 as a demand for those matrices appeared during the method development study. The slightly 360 modified procedure described in the sample preparation section was used. None of the 361 targeted pesticides was detected in all bee samples, whereas 26 wax samples showed 362 significant levels of Flusilazole, 17 presented positive results for Rotenone, Pirimicarb has 363 been found in 10 samples, 4 and 3 samples showed traces of Bitertanol and Atrazine, 364 respectively. Only one wax revealed levels in 2-Hydroxytertbutylazine. These results 365 demonstrated the flexibility of the procedure regarding matrix types. The range of chemicals analyzed can be extended while keeping the same extraction method but combining LC-366 367 MS/MS and GC-MS/MS. This approach has further been recently developed in our laboratory 368 and pesticides such as Coumaphos, Bromopropylate, Vinclozine, tau-Fluvalinate and 369 Lindane, among others, can now additionally be detected in honey, bee or wax samples (data 370 not shown).

371

4. CONCLUSIONS

372 A rapid, reliable, time and resource saving analytical method is reported for the measurement of a wide range of different chemicals used in apiculture or in the surrounding agriculture in 373 374 the context of a bee mortality study. The multi-residue analytical procedure developed in this 375 study was based on an on-column liquid-liquid extraction step using diatomaceous earth as 376 inert solid support. Extracts were analyzed without further purification by LC-MS/MS in ESI 377 mode. Extraction by OCLLE using the commercially available Chem Elut cartridges has proven to be efficient for a wide range of pesticides, nearly independently of their polarity. 378 379 The use of LC-MS/MS permitted to avoid undesirable derivatization steps while lowering 380 sample clean-up requirements, compared to GC-MS. The extraction and purification have then be considerably reduced and simplified. The MRM allowed to separate the 17 target 381 pesticides in less than 15 minutes with good specificity. 382

383 A complete validation following the European Commission decision 2002/657/EC dedicated 384 to some residues in living animals or their derived products has been performed for the 17 pesticides belonging to widely chemically different families, from organophosphorous to 385 386 triazines, including ureas, carbamates, pyrazoles, nicotinoids or pyrimidines. Specificity, 387 calibration curves, trueness, reproducibility, sensitivity and robustness have been tested 388 successfully, demonstrating the suitability of this method for selected compounds. The list of 389 pesticides can easily be extended by adding a GC-MS/MS injection of the extracts. Other 390 matrices such as wax and bee have also been included in the study by slightly adapting the 391 extraction procedure.

392

393

394 Acknowledgements

- 395 The authors thank the Fonds Européen de Développement Régional (FEDER), the Fonds
- 396 Social Européen (FSE) and the Service Public Fédéral Santé Publique (Formerly Ministère
- des Classes Moyennes et de l'Agriculture DG6) for their financial support.
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399	References
400	
401	[1] M. Fernández, Y. Picó, J. Mañes, J. Food Protect. 65 (2002) 1502.
402	[2] J.J. Jiménez, J. L. Bernal, M.J. del Nozal, M.T. Martín, A.L. Mayorga, J. Chromatogr. A
403	829 (1998) 269.
404	[3] B. Albero, C. Sanchez-Brunete, J.L. Tadeo, J. A.O.A.C. Int. 84 (2001) 1165.
405	[4] B. Albero, C. Sanchez-Brunete, J.L. Tadeo, J. Agric. Food Chem. 52 (2004) 5828.
406	[5] S.R. Rissato, M.S. Galhiane, F.R.N. Knoll, B.M. Apon, J. Chromatogr. A 1048 (2004)
407	153.
408	[6] F.E. Ahmed, Trend. Anal. Chem. 20 (2001) 649.
409	[7] C. Blasco, C.M. Lino, Y. Picó, A. Pena, G. Font, M.I.N. Silveira, J. Chromatogr. A 1049
410	(2004) 155.
411	[8] Y.R. Tahboub, M.F. Zaater, T.A. Barri, Anal. Chim. Acta 558 (2006) 62.

- 412 [9] S.A. Barker, J. Chromatogr. A 880 (2000) 63.
- 413 [10] E.M. Kristenson, L. Ramos, U.A.T. Brinkman, Trend. Anal. Chem. 25 (2006) 96.
- 414 [11] B. Albero, C. Sanchez-Brunete, J.L. Tadeo, Talanta 66 (2005) 917.
- 415 [12] N. Campillo, R. Peñalver, N Aguinaga, M. Hernández-Córdoba, Anal. Chim. Acta 562416 (2006) 9.
- 417 [13] N. Ochiai, K. Sasamoto, H. Kanda, S. Nakamura, J. Chromatogr. A 1130 (2006) 83.
- 418 [14] Commission Decision 2002/657/EC, Off. J. Eur. Commun., 12 August 2002.
- 419 [15] J. Klein, L. Alder, A. Schreiber, Application note from Applied Biosystem TM available
- 420 at http://docs.appliedbiosystems.com/pebiodocs/00113222.pdf.
- 421 [16] K.R Engh. In: Kroschwitz, J.I. (Ed.), Kirk-Othmer 1993. Diatomite.
- 422 [17] J.F.Lemonas, Am. Ceramic Soc. Bull. 76 (1997) 92.

- 423 [18] S. Fustinoni, L. Campo, C. Colosio, S. Birindelli, V. Foà, J. Chromatogr. B 814 (2005)
- 424 251.
- 425 [19] B. Büchele, W. Zugmaier, F. Genze, T. Simmet, J. Chromatogr. B 829 (2005) 144.
- 426 [20] F. Sicbaldi, A. Sarra, G. L. Copeta, J. Chromatogr. A 765 (1997) 23.
- 427 [21] F. Le Floch, M. T. Tena, A. Ríos, M. Valcárcel., Talanta 46 (1998) 1123.
- 428 [22] D. Perret, A. Gentili, S. Marchese, M. Sergi, G. D'Ascenzo, J. A.O.A.C. Int. 85 (2002)
- 429 724.
- 430 [23] C. Sánchez de la Torre, M. A. Martínez, E. Almarza, Forensic Sci. Int. 155 (2005) 193.
- 431 [24] P. Fidente, S. Seccia, F. Vanni, P. Morrica, J. Chromatogr. A 1094 (2005) 175.
- 432 [25] U. N. Andersen, G. Seeber, D. Fiedler, K. N. Raymond, D. Lin, D. Harris, J. Am. Soc.
- 433 Mass Spectr. 17 (2006) 292.
- 434 [26] I. Clarot, A. Regazzeti, N. Auzeil, F. Laadani, M. Citton, P. Netter, A. Nicolas. J,
 435 Chromatogr. A 1087 (2005) 236.
- 436 [27] G. Chiti, M. Municchi, V. Paschetta, D. Nistri, G. Roncucci, J. Chromatogr. B 809
 437 (2004) 167.
- 438 [28] Commission Recommendation 1999/333/EC, Off. J. Eur. Commun., 3 March 1999.
- 439 [29] Council Directive 96/23/EC, Off. J. Eur. Commun., 29 April 1996.

Pesticide name	Abbreviation used	State	Purity	Level
			(%)	$(ng mL^{-1})$
Amidosulfuron	Am	solid	97.5	0.4
Atrazine	At	solution	99.5	0.4
Carbofuran	Ca	solid	99.5	0.4
Chlorotoluron	Ch	solution	99	20.0
Diethofencarb	De	solid	97.5	2.0
Dimethoate	Dm	solution	99	2.0
Fipronil	Fi	solid	96.5	10.0
Imidacloprid	Im	solution	97	2.0
Isoxaflutole	Is	solution	98.5	2.0
Linuron	Li	solution	99.5	2.0
Methiocarb	Mh	solution	98.5	10.0
Methiocarb sulfoxide	MhS	solution	96	20.0
Metosulam	Мо	solid	99.5	2.0
Pirimicarb	Pi	solution	98	0.4
Rimsulfuron	Ri	solid	99.5	0.4
Simazine	Si	solid	98	2.0
2-Hydroxyterbuthylazine	ТОН	solution	98.5	1.0

Table 1: List of the pesticide abbreviated names used all along this manuscript as well as method development concentration levels in the surrogate standard solution.

Table 2: List	of acquisition	parameters.	Masses	in bol	d underlined	are	those	used	for
quantification.									

Windows		Compounds	Precurseur	Product	Dwell times	Collision
	time (min)		ions (m/z)	ions (m/z)	(s)	voltages
1	5.84	Methiocarb sulfoxide	242	170	0.2	20
			242	<u>185</u>	0.2	10
	6.16	Imidacloprid	256	175	0.2	15
			256	<u>209</u>	0.2	15
	6.18	2-Hydroxytertbutylazine	212	114	0.1	20
			212	<u>156</u>	0.1	15
	6.38	Dimethoate	230	171	0.1	15
			230	<u>199</u>	0.1	10
2	8.26	Pirimicarb	239	<u>182</u>	0.2	15
			239	195	0.2	10
	8.68	Simazine	202	<u>124</u>	0.2	15
	0.00		202	132	0.2	15
3	9.57	Carbofuran	222	123	0.2	15
5	9.51	Carboruran	222	<u>165</u>	0.2	10
	9.93	Amidosulfuron	370	218	0.2	20
	9.95	Annuosunuion	370	<u>218</u>	0.2	20 10
	10.09	Chlorotoluron	213	$\frac{201}{140}$	0.2	20
	10.09	Ciliolololuloli	213		0.2	20 15
	10.11	Dimaulfunan		<u>168</u> 182		13 20
	10.11	Rimsulfuron	432	<u>182</u> 325	0.2	
	10.27	A 4	432		0.2	15
	10.27	Atrazine	216	146	0.2	20
	10 (1		216	<u>174</u>	0.2	15
	10.61	Metosulam	419	<u>175</u>	0.1	20
			419	228	0.1	15
4	12.11	Methiocarb	226	121	0.2	15
			226	<u>169</u>	0.2	10
	12.25	Diethofencarb	268	180	0.1	15
			268	<u>226</u>	0.1	10
	12.29	Linuron	249	<u>160</u>	0.2	15
			249	182	0.2	15
	12.29	Linuron D6	256	<u>161</u>	0.1	15
	12.62	Isoxaflutole	360	<u>251</u>	0.2	10
			360	262	0.2	10
5	13.47	Flusilazole	316	165	0.2	25
			316	247	0.2	15
	13.68	Bitertanol	338	<u>99</u>	0.2	15
			338	269	0.2	5
	13.81	Rotenone	395	192	0.2	20
	- • • -		395	<u>213</u>	0.2	20
6	14.55	Fipronil	437	290	0.2	25
0	17.33	1 ipionin	437	<u>368</u>	0.2	23 15

Table 3: Validation process data showing the concentration range inside which the linearity was tested, levels (ng g⁻¹ honey) of the 3 QC samples analyzed in 6 replicates, and RSD obtained for reproducibility test (QC level #1 and #3 analyzed in 6 replicates at 3 distinct days by 3 different operators).

	Linearity			QC levels			(n = 18)	Reproducibility
	Levels	R ²	# 1	#2	#3	Recovery	RSD	RSD
	$(ng g^{-1})$		$(ng g^{-1})$	$(ng g^{-1})$	$(ng g^{-1})$	Mean %		n = 36
Amidosulfuron	[0.1-0.4]	0.995	0.1	0.3	0.4	79	9	13
Atrazine	[0.1-0.4]	0.994	0.1	0.3	0.4	81	14	24
Carbofuran	[0.1-0.4]	0.977	0.1	0.3	0.4	89	12	15
Chlorotoluron	[5-20]	0.999	5.0	15	20	90	13	18
Diethofencarb	[0.5-2]	0.982	0.5	1.5	2.0	80	11	18
Dimethoate	[0.5-2]	0.999	0.5	1.5	2.0	89	11	13
Fipronil	[2.5-10]	0.979	2.5	7.5	10	72	12	20
Imidacloprid	[0.5-2]	0.999	0.5	1.5	2.0	86	10	14
Isoxaflutole	[0.5-2]	0.961	0.5	1.5	2.0	87	10	12
Linuron	[0.5-2]	0.961	0.5	1.5	2.0	79	10	27
Methiocarb	[2.5-10]	0.994	2.5	7.5	10	78	15	22
Methiocarb sulfoxide	[5-20]	0.999	5.0	15	20	89	11	11
Metosulam	[0.5-2]	0.986	0.5	1.5	2.0	83	16	19
Pirimicarb	[0.1-0.4]	0.994	0.1	0.3	0.4	88	7	21
Rimsulfuron	[0.1-0.4]	0.921	0.1	0.3	0.4	71	12	24
Simazine	[0.5-2]	0.985	0.5	1.5	2.0	82	9	18
2-Hydroxyterbuthylazine	[0.25-1]	0.996	0.25	0.75	1.0	78	9	8

Table 4: Number of the real honey samples in which pesticides where detected and measured levels (ng g⁻¹ honey). Decision limits and detection capacity (CC α and CC β) are expressed in ng g⁻¹ of raw honey. ND = non-detected.

Pesticides	Number of positive –	Level rang	$ge(ng g^{-1})$	CCα	ССβ		
	samples	min max		$(ng g^{-1})$	$(ng g^{-1})$		
Rotenone	22	<0.5	>2	0.057	0.069		
Flusilazole	8	< 0.05	< 0.05	0.093	0.141		
Methiocarb sulfoxide	8	0.09	0.31	0.0002	0.0002		
Imidacloprid	5	<ccβ< td=""><td><ccβ< td=""><td>0.069</td><td>0.084</td></ccβ<></td></ccβ<>	<ccβ< td=""><td>0.069</td><td>0.084</td></ccβ<>	0.069	0.084		
Bitertanol	1	0.12	0.12	0.015	0.018		
Carbofuran	1	> 0.6	> 0.6	0.253	0.317		
Pirimicarb	1	<ccβ< td=""><td><ccβ< td=""><td>0.053</td><td>0.071</td></ccβ<></td></ccβ<>	<ccβ< td=""><td>0.053</td><td>0.071</td></ccβ<>	0.053	0.071		
Amidosulfuron	0	ND	ND	0.089	0.108		
Atrazine	0	ND	ND	0.197	0.275		
Chlorotoluron	0	ND	ND	0.003	0.004		
Fipronil	0	ND	ND	0.011	0.015		
Isoxaflutole	0	ND	ND	0.172	0.204		
Linuron	0	ND	ND	0.321	0.462		
Methiocarb	0	ND	ND	0.011	0.015		
Rimsulfuron	0	ND	ND	0.670	0.930		
Simazine	0	ND	ND	0.250	0.326		
Diethofencarb	0	ND	ND	0.579	0.751		
Dimethoate	0	ND	ND	0.060	0.073		
Metosulam	0	ND	ND	0.943	1.232		
2-Hydroxytertbutylazine	0	ND	ND	0.0003	0.0003		

Figure 1: Comparison betwween recoveries obtained by OCLLE and classical LLE

Figure 2: Typical reconstructed chromatogram based on specific masses (in bold underlined in Table 1) obtained for a methanolic standard solution using the Polaris C18-A column and the optimized LC conditions.

Figure 3: Collision voltage optimization for Imidacloprid. The retained voltage (15 V) produced the most intense signals for both transitions yielding to masses 175 and 209.





